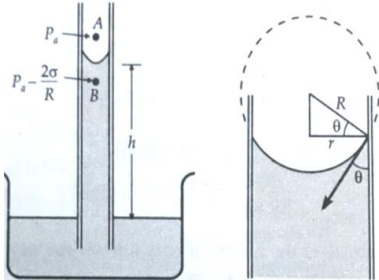
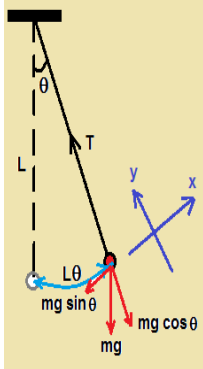
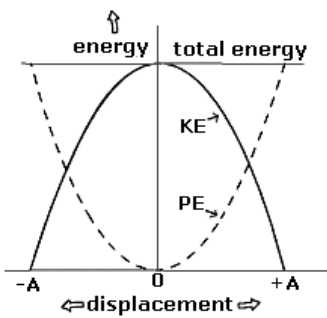


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| | <p>concave meniscus</p> <p>Due to this excess pressure, the liquid rises</p> <p>In the capillary tube to height h, $p \frac{r}{R} = \cos \theta$; $R = \frac{r}{\cos \theta}$;</p> <p>$p = \frac{2\sigma \cos \theta}{r} = h\rho g$</p> <p>$h\rho g = \frac{2\sigma \cos \theta}{r}$; $h = \frac{2\sigma \cos \theta}{r \rho g}$</p>  | <p>1/2</p> <p>1/2</p> <p>1</p> <p>1/2 dig.</p> |
| 7. | <p>(a) According to first law of thermodynamics: - The change in the internal energy of a closed system is equal to the amount of heat supplied to the system, minus the amount of work done by the system on its surroundings.</p> <p>$\Delta Q = \Delta U + \Delta W$</p> <p>Where:</p> <p>$\Delta Q$ is the heat supplied to the system by the surroundings</p> <p>ΔW is the work done by the system by the surroundings</p> <p>ΔU is the change in internal energy of the system.</p> <p>(b) Internal energy is sum of KE and PE of particles of thermodynamic system</p> | <p>1</p> <p>1</p> <p>1</p> |
| 8. | <p>(a) The number of independent ways in which a molecule of gas can move is called the degree of freedom.</p> <p>(b) (i) Monoatomic gas is made of a single atom.</p> <p>i.e., $N = 1$, so $K = 0$, therefore $f = 3 \times 1 - 0 = 3$.</p> <p>Degrees of freedom of monoatomic gas molecule is 3.</p> <p>(ii) Diatomic gas molecule is made of two atoms.</p> <p>i.e., $N = 2$, So $K = 1$,</p> <p>Degrees of freedom, $f = 3N - K$,</p> <p>$f = 3 \times 2 - 1 = 5$</p> | <p>1</p> <p>1</p> <p>1</p> |
| 9. | <p>$PV = nRT$</p> <p>$V = \frac{nRT}{P}$</p> <p>For one mole of a gas at STP we have</p> <p>$V = \frac{1 \times 8.314 \times 273}{1.013 \times 10^5}$</p> <p>$V = 0.0224 m^3$</p> <p>$V = 22.4 \text{ litres}$</p> | <p>1</p> <p>1</p> <p>1</p> |

| | | |
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| 10. | <p>From pascal's law</p> $P_1 = P_2$ $\frac{F_1}{A_1} = \frac{F_2}{A_2}$ $\frac{F_1}{\pi r_1^2} = \frac{F_2}{\pi r_2^2}$ $F_1 = \frac{F_2 r_1^2}{r_2^2}$ $F_1 = \frac{1350 \times 9.8 \times (5 \times 10^{-2})^2}{(15 \times 10^{-2})^2}$ $F_1 = 1470 \text{ N}$ $F_1 = 1.47 \times 10^3 \text{ N}$ <p>$P_1 = F_1 / A_1$ $P_1 = 1.9 \times 10^5 \text{ Pa}$</p> <p>(OR)</p> <p>$H = 2T \cos \theta / r \sin \theta$ Substitution of values $= 2.8 \times 10^{-2} \text{ N / m}$</p> | <p>$\frac{1}{2}$</p> <p>$\frac{1}{2}$</p> <p>$\frac{1}{2}$</p> <p>$\frac{1}{2}$</p> <p>$\frac{1}{2}$</p> <p>$\frac{1}{2}$</p> <p>1</p> <p>1</p> <p>1</p> |
| 11. | <p>A motion be Simple harmonic motion only when,</p> <ol style="list-style-type: none"> 1. Acceleration of particle is just opposite to motion of body 2. Acceleration is directly proportional to displacement e.g., $a = -\omega^2 x$  <p>restoring force, $F = -mg \sin \theta$,</p> <p>When displacement of pendulum is very small , then $\sin \theta \approx \theta$</p> <p>so, $F = -mg \theta$, also here it is clear , $\theta = x/L$</p> <p>$\therefore F = -mg x/L$</p> <p>Now use $F = ma$ { Newton's second law }</p> <p>$ma = -mg x/L \Rightarrow a = -g x/L$</p> <p>Now, compare both the expressions ,</p> <p>$\therefore \omega^2 = g/L$</p> <p>we know, $\omega = 2\pi/T$, here T is time period .</p> | <p>1 dig.</p> <p>$\frac{1}{2}$</p> <p>$\frac{1}{2}$</p> <p>$\frac{1}{2}$</p> |

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| | | <p>so, $\{2\pi/T\}^2 = g/L$ $\Rightarrow T = 2\pi\sqrt{L/g}$ Hence, for pendulum time period is $T = 2\pi\sqrt{L/g}$</p> <p style="text-align: center;">(OR)</p> <p>Derivation for total energy of the particle executing simple harmonic motion.</p> <p>Expression for KE</p> <p>Expression for PE</p> <p>The variation of kinetic energy and potential energy with displacement.</p> <div></div> | <p>1/2</p> <p>1/2</p> <p>1 1/2</p> <p>1</p> |
| | | SECTION C | 1 x 5 = 5 |
| 12 | <p>(i) B (ii) C (iii) D (iv) C and D (any one) (v) D</p> | | |
| | | THE END OF MARKING SCHEME | |